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## Ub-FANCD2(557-565)-FP (human, synthetic)

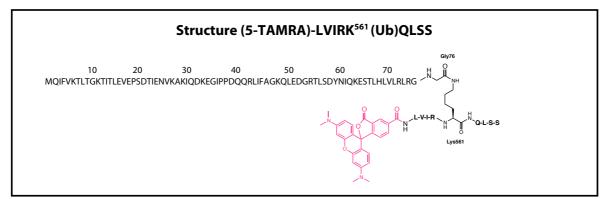
UbiQ code	:	UbiQ-029
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Protocol # : P01082013-001

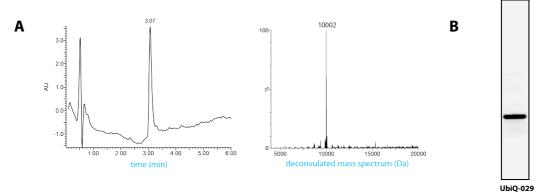
## **Product Information**

Amount	: 0.16 mg, lyophilized powder (purple)	
Purity	: ≥95% by RP-HPLC and SDS-PAGE	
Mol. Weight	: 10002 Da by MS (calc Mw 10002 Da)	
Storage	: -20°C; buffered solution at -80°C. Please avoid multiple freeze/thaw cycles	5.

**Background.** Class II fluorescence polarization HTS reagent based on the peptide sequence 557 – 565 of FANCD2, a DNA repair protein which is monoubiquitinated on Lys561. The peptide is modified on the N-terminus with a 5-carboxytetramethylrhodamine (5-TAMRA) and conjugated at Lys561 to ubiquitin via a native isopeptide bond. Typical substrate concentrations range from 10–100 nM. DUB concentrations can range from 0.01-10 nM but depend on specific assay conditions and method of detection.



**Important:** sample preparation. Dissolve the powder in as little DMSO as possible (*e.g.* 10 mg/mL = 1000  $\mu$ M, if not clear then heat slightly) and <u>add this DMSO stock</u> slowly to the required buffer (please note the order of addition). *If possible, it is prefered to dilute the DMSO stock into milliQ and then buffer the solution.* For experimental details see page 2 and the open-access reference 4. The concentration of 5-TAMRA-Lys(Ub)-Gly-OH can be verified by comparing the fluorescence intensity with that of a known concentration of TAMRA.



**A: LC-MS analysis.** Mobile phase A = 1% CH<sub>3</sub>CN, 0.1% formic acid in water (milliQ) and B = 1% water (milliQ) and 0.1% formic acid in CH<sub>3</sub>CN. Phenomenex Kinetex C18, (2.1×50 mm, 2.6  $\mu$ M); flow rate = 0.6 mL/min, runtime = 6 min, column T = 40°C. Gradient: 5%  $\Rightarrow$  95% over 3.5 min. **B: SDS-PAGE analysis.** Fluorescence scan (exc 550 nm, emi 590 nm), 12% SDS-PAGE gel.



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## **Experimental Details Fluorescence polarization assay**

FP assays were performed on a PerkinElmer Wallac EnVision 2010 Multilabel Reader with a 531 nm excitation filter and two 579 nm emission filters. Fluorescence intensities were measured in the S (parallel) and P (perpendicular) direction. FP values are given in mP (millipolarization) and calculated using the following formula:

$$Polarization (mP) = \frac{S - (G \cdot P)}{S + (G \cdot P)} \cdot 1000$$

The confocal optics were adjusted with the average P and S values for TAMRA and the grating factor (G) was determined using a polarization value (L) for TAMRA of 50 mP using the following formula:

$$G = \frac{average\ S}{average\ P} \cdot \frac{1 - (\frac{L}{1000})}{1 + (\frac{L}{1000})}$$

The assays were performed in "non binding surface flat bottom low flange" black 384-well plates (Corning) at room temperature in a buffer containing 20 mM Tris·HCl, pH 7.5, 5 mM DTT, 100 mM NaCl, 1 mg/mL 3-[(3-cholamidopropyl) dimethylammonio] propanesulfonic acid (CHAPS) and 0.5 mg/mL bovine gamma globulin (BGG). Each well had a volume of 20  $\mu$ L. Buffer and enzyme were predispensed and the reaction was started by the addition of substrate. Kinetic data was collected in intervals of 2.5 or 3 min. From the obtained polarization values (*P*<sub>t</sub>) the amount of processed substrate (*S*<sub>t</sub>) was calculated with to the following equation:

$$S_t = S_0 - \left[ S_0 \times \frac{P_t - P_{min}}{P_{max} - P_{min}} \right]$$

Where  $P_t$  is the polarization measured (in mP);  $P_{max}$  is the polarization of 100% unprocessed substrate (determined for every reagent at all used substrate concentrations);  $P_{min}$  is the polarization of 100% processed substrate;  $S_0$  is the amount of substrate added to the reaction.

From the obtained  $P_t$  values the values for initial velocities ( $v_i$ ) were calculated, which were used to determine the Michaelis-Menten constants ( $K_m$ ,  $V_{max}$  and  $k_{cat}$ ) by fitting the data according to the formula below (where  $k_{cat} = V_{max}/[E]$ ). All experimental data was processed using Ms Excel and Prism 4.03 (GraphPad Software, Inc.).

$$v_i = \frac{V_{max} \times S_0}{K_m + S_0}$$

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Literature. (1) Tirat, A. et al. Anal. Biochem. 2005, 343, 244-255. (2) Huang et al. Methods Mol. Biol. 2009, 565, 127. (3) Levine et al. Anal. Biochem. 1997, 247, 83. (4) Geurink and El Oualid et al. ChemBiochem, 2012, 13, 293. (5) Sheng et al. Nat. Struct. Mol. Biol. 2006, 13, 285.